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Two Isoxazolidines

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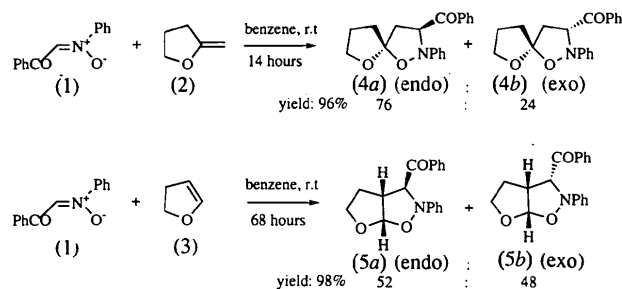
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Abstract

The crystal structures of *trans*-phenyl 2-phenyl-1,6-dioxo-2-azaspiro[4.4]non-3-yl ketone, C₁₉H₁₉NO₃, and (3 α ,3 α ,6 α)-hexahydro-2-phenylfuro[3,2-*d*]isoxazol-3-yl phenyl ketone, C₁₈H₁₇NO₃, are reported. In both compounds, the isoxazolidine rings adopt envelope conformations in which the O atom is bent out of the approximate plane of the other four ring atoms. Modest to negligible *endo* selectivities were confirmed in 1,3-dipolar cycloadditions of benzoylmethylenedianiline *N*-oxide with enol ethers.

Comment

The structures of the two title molecules, *trans*-phenyl(2-phenyl-1,6-dioxo-2-azaspiro[4.4]non-3-yl)methanone, (4*a*), and (3 α ,3 α ,6 α)-(hexahydro-2-phenylfuro[3,2-*d*]isoxazol-3-yl)phenylmethanone, (5*a*), are shown in Figs. 1 and 2, respectively. The isoxazolidine rings in



both molecules adopt envelope conformations in which the O atoms (O3) are bent out of plane. These structures confirm that both of the major cycloadducts, (4*a*) and (5*a*), are formed by *endo* cycloaddition during the 1,3-dipolar cycloaddition. The *endo* adducts have the O atom of compound (2) or the ring of compound (3) near the N atom of the planar nitron in the transition state. The X-ray structures of other related isoxazolidines have been reported by Lizamma, Varghese & Sankararaman (1993) and Bravo, Bruche, Farina, Fronza, Meille & Merli (1993).

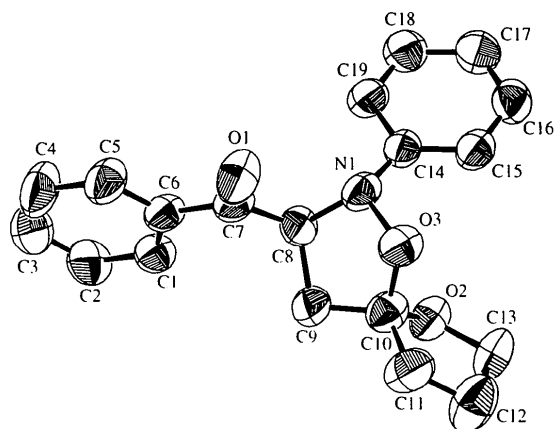


Fig. 1. An ORTEP (Johnson, 1965) view of compound (4*a*) with the atomic labeling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are omitted for clarity.

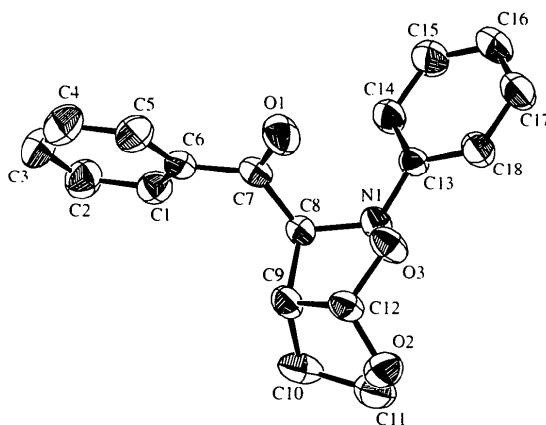


Fig. 2. An ORTEP (Johnson, 1965) view of compound (5*a*) with the atomic labeling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are omitted for clarity.

Experimental

Nitron (1) (benzoylmethylenedianiline *N*-oxide) was reacted with olefins (2) (tetrahydro-2-methylenefuran) and (3) (2,3-dihydrofuran) as shown in the scheme above. Nitron (1) was prepared according to the literature method of Avera, Cum, Strigno d'Alcontres & Uccella (1972). The products were separated by chromatography on silica gel. Compound (4*a*) was recrystallized from acetonitrile several times to afford single crystals (m.p. 372–373 K) suitable for X-ray diffraction analysis. Compound (5*a*) was recrystallized from a mixture of hexane and ethyl acetate. Compounds (4*a*) and (5*a*) have been reported previously in the literature (Fisera, Dandarova, Kovac, Gaplovsk'y, Patuš & Goljer, 1982), but their stereochemistries were assigned based on the coupling constants of ¹H NMR spectra. The major products from both reactions [*i.e.* compounds (4*a*) and (5*a*)] were examined crystallographically to confirm the assigned stereochemistries.

Compound (4a)*Crystal data*C₁₉H₁₉NO₃ $M_r = 309.36$

Monoclinic

 $P2_1/n$ $a = 8.678$ (7) Å $b = 16.19$ (1) Å $c = 11.53$ (1) Å $\beta = 99.00$ (1)° $V = 1599$ (2) Å³ $Z = 4$ $D_x = 1.285$ Mg m⁻³ D_m not measured*Data collection*Syntex (Crystal Logic)
diffractometer $\theta/2\theta$ scansAbsorption correction:
none

2452 measured reflections

2140 independent reflections

1016 observed reflections

 $[I > 3\sigma(I)]$ *Refinement*Refinement on F^2 $R = 0.046$ $wR = 0.051$ $S = 1.457$

1016 reflections

209 parameters

H atoms were placed in
calculated positions $w = 1/\sigma^2(F_o)$ $(\Delta/\sigma)_{\max} = 0.002$ Cu $K\alpha$ radiation $\lambda = 1.5418$ ÅCell parameters from 22
reflections $\theta = 8.2$ – 13.2 ° $\mu = 0.664$ mm⁻¹ $T = 298$ K

Irregular

 $0.25 \times 0.20 \times 0.20$ mm

Colorless

 $R_{\text{int}} = 0.02$ $\theta_{\max} = 57.5$ ° $h = 0 \rightarrow 9$ $k = 0 \rightarrow 17$ $l = -12 \rightarrow 12$

3 standard reflections

monitored every 97

reflections

intensity decay: 0.1%

 $\Delta\rho_{\max} = 0.28$ e Å⁻³ $\Delta\rho_{\min} = -0.36$ e Å⁻³

Extinction correction:

Larson (1967) isotropic

Extinction coefficient:

2468 (340)

Atomic scattering factors

from *International Tables*
for *X-ray Crystallography*

(1974, Vol. IV, Tables

2.2B and 2.3.1)

C14	-0.0706 (7)	0.3344 (3)	-0.0858 (4)	0.057 (4)
C15	-0.1228 (8)	0.2927 (3)	0.0061 (4)	0.066 (4)
C16	-0.2806 (8)	0.2793 (3)	0.0026 (5)	0.073 (5)
C17	-0.3892 (7)	0.3075 (3)	-0.0891 (5)	0.076 (5)
C18	-0.3357 (7)	0.3478 (3)	-0.1807 (5)	0.076 (5)
C19	-0.1795 (7)	0.3611 (3)	-0.1799 (4)	0.070 (4)

Table 2. Selected geometric parameters (Å, °) for (4a)

N1—C14	1.402 (6)	C6—C7	1.479 (6)
N1—O3	1.444 (4)	C7—C8	1.521 (6)
N1—C8	1.474 (5)	C8—C9	1.543 (6)
O1—C7	1.211 (5)	C9—C10	1.503 (6)
O2—C10	1.408 (5)	C10—C11	1.519 (6)
O2—C13	1.434 (5)	C11—C12	1.480 (7)
O3—C10	1.434 (5)	C12—C13	1.487 (7)
C1—C2	1.383 (6)	C14—C15	1.390 (6)
C1—C6	1.390 (6)	C14—C19	1.391 (6)
C2—C3	1.373 (7)	C15—C16	1.381 (7)
C3—C4	1.378 (7)	C16—C17	1.379 (6)
C4—C5	1.383 (6)	C17—C18	1.382 (7)
C5—C6	1.391 (6)	C18—C19	1.371 (7)
C14—N1—O3	113.9 (4)	C10—C9—C8	104.1 (4)
C14—N1—C8	118.9 (4)	O2—C10—O3	110.4 (4)
O3—N1—C8	106.7 (4)	O2—C10—C9	110.3 (4)
C10—O2—C13	107.7 (4)	O2—C10—C11	106.7 (4)
C10—O3—N1	107.4 (3)	O3—C10—C9	103.9 (4)
C2—C1—C6	121.2 (5)	O3—C10—C11	106.4 (4)
C3—C2—C1	119.4 (5)	C9—C10—C11	118.8 (5)
C2—C3—C4	120.5 (5)	C12—C11—C10	105.4 (5)
C3—C4—C5	120.2 (5)	C11—C12—C13	105.4 (5)
C4—C5—C6	120.2 (5)	O2—C13—C12	105.6 (4)
C1—C6—C5	118.6 (4)	C15—C14—C19	118.8 (5)
C1—C6—C7	123.1 (4)	C15—C14—N1	123.4 (5)
C5—C6—C7	118.4 (4)	C19—C14—N1	117.6 (5)
O1—C7—C6	121.0 (5)	C16—C15—C14	119.6 (5)
O1—C7—C8	120.9 (4)	C17—C16—C15	121.7 (5)
C6—C7—C8	118.1 (4)	C16—C17—C18	118.1 (5)
N1—C8—C7	109.8 (4)	C19—C18—C17	121.3 (5)
N1—C8—C9	105.4 (4)	C18—C19—C14	120.4 (5)
C7—C8—C9	110.0 (4)		
O3—N1—C8—C7	108.6 (4)	C8—C9—C10—O2	-89.4 (5)
O3—N1—C8—C9	-9.8 (5)	C8—C9—C10—O3	28.9 (5)
O3—N1—C14—C15	-10.3 (7)	C8—C9—C10—C11	147.0 (5)
O3—N1—C14—C19	175.2 (4)	C13—O2—C10—O3	90.4 (5)
C8—N1—C14—C15	-137.4 (5)	C13—O2—C10—C9	-155.3 (4)
C8—N1—C14—C19	48.1 (6)	C13—O2—C10—C11	-24.8 (5)
C10—O2—C13—C12	31.6 (5)	O2—C10—C11—C12	8.4 (6)
C8—N1—O3—C10	29.3 (5)	O3—C10—C11—C12	-109.5 (5)
C14—N1—O3—C10	-103.9 (5)	C9—C10—C11—C12	133.8 (5)
N1—O3—C10—O2	81.8 (4)	C10—C11—C12—C13	10.3 (6)
N1—O3—C10—C9	-36.5 (5)	C11—C12—C13—O2	-25.3 (6)
N1—O3—C10—C11	-162.8 (4)	N1—C14—C15—C16	-174.9 (5)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²) for (4a)

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j.$$

	<i>x</i>	<i>y</i>	<i>z</i>	U_{eq}
N1	0.0877 (6)	0.3450 (2)	-0.0940 (3)	0.063 (3)
O1	0.2585 (5)	0.3493 (2)	-0.2700 (3)	0.091 (3)
O2	0.1376 (4)	0.4269 (2)	0.1370 (3)	0.067 (3)
O3	0.1923 (4)	0.3218 (2)	0.0110 (3)	0.069 (3)
C1	0.2090 (5)	0.5703 (3)	-0.2737 (4)	0.062 (4)
C2	0.2278 (6)	0.6383 (3)	-0.3428 (5)	0.075 (4)
C3	0.2665 (7)	0.6267 (4)	-0.4527 (5)	0.089 (5)
C4	0.2900 (8)	0.5483 (4)	-0.4930 (5)	0.101 (5)
C5	0.2727 (7)	0.4803 (3)	-0.4236 (4)	0.083 (5)
C6	0.2307 (6)	0.4906 (3)	-0.3129 (4)	0.061 (4)
C7	0.2144 (6)	0.4162 (3)	-0.2414 (4)	0.062 (4)
C8	0.1410 (6)	0.4260 (3)	-0.1307 (4)	0.060 (4)
C9	0.2649 (6)	0.4552 (3)	-0.0281 (4)	0.077 (4)
C10	0.2478 (6)	0.3965 (3)	0.0701 (4)	0.066 (4)
C11	0.3919 (6)	0.3732 (4)	0.1565 (5)	0.089 (5)
C12	0.3452 (8)	0.3806 (4)	0.2742 (5)	0.103 (5)
C13	0.1730 (7)	0.3917 (3)	0.2523 (4)	0.083 (5)

Compound (5a)*Crystal data*C₁₈H₁₇NO₃ $M_r = 295.34$

Orthorhombic

 $Pbca$ $a = 10.181$ (4) Å $b = 28.90$ (1) Å $c = 10.071$ (2) Å $V = 2963$ (2) Å³ $Z = 8$ $D_x = 1.324$ Mg m⁻³ D_m not measuredCu $K\alpha$ radiation $\lambda = 1.5418$ ÅCell parameters from 20
reflections $\theta = 9.7$ – 17.3 ° $\mu = 0.694$ mm⁻¹ $T = 298$ K

Needle

 $0.15 \times 0.10 \times 0.05$ mm

Pale yellow

Data collection

Rigaku AFC-5R diffractometer	$\theta_{\max} = 59.99^\circ$
$\theta/2\theta$ scans	$h = 0 \rightarrow 11$
Absorption correction: none	$k = 0 \rightarrow 32$
2560 measured reflections	$l = 0 \rightarrow 11$
2560 independent reflections	3 standard reflections
1199 observed reflections	monitored every 150 reflections
$[I > 3\sigma(I)]$	intensity decay: 0.1%

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\max} = 0.010$
$R = 0.048$	$\Delta\rho_{\max} = 0.42 \text{ e } \text{\AA}^{-3}$
$wR = 0.059$	$\Delta\rho_{\min} = -0.56 \text{ e } \text{\AA}^{-3}$
$S = 1.337$	Extinction correction: none
1199 reflections	Atomic scattering factors
199 parameters	from <i>International Tables</i>
H atoms placed in calculated positions	for <i>X-ray Crystallography</i>
$w = 1/\sigma^2(F_o)$	(1974, Vol. IV, Tables 2.2B and 2.3.1)

C12—O2—C11	109.1 (3)	C7—C8—C9	109.6 (3)
C12—O3—N1	105.1 (2)	C10—C9—C12	103.4 (3)
C13—N1—C8	123.3 (3)	C10—C9—C8	113.9 (3)
C13—N1—O3	109.9 (3)	C12—C9—C8	103.3 (3)
C8—N1—O3	104.6 (3)	C11—C10—C9	104.1 (3)
C6—C1—C2	119.4 (4)	O2—C11—C10	104.7 (3)
C3—C2—C1	120.3 (4)	O2—C12—O3	112.0 (3)
C2—C3—C4	120.5 (4)	O2—C12—C9	107.5 (3)
C3—C4—C5	119.6 (4)	O3—C12—C9	106.4 (3)
C4—C5—C6	120.5 (4)	C18—C13—C14	119.9 (3)
C5—C6—C1	119.6 (4)	C18—C13—N1	120.2 (3)
C5—C6—C7	117.9 (4)	C14—C13—N1	119.5 (3)
C1—C6—C7	122.5 (4)	C15—C14—C13	119.3 (4)
O1—C7—C6	120.8 (4)	C16—C15—C14	120.8 (4)
O1—C7—C8	120.4 (3)	C15—C16—C17	119.3 (4)
C6—C7—C8	118.7 (3)	C16—C17—C18	120.7 (4)
N1—C8—C7	115.3 (3)	C13—C18—C17	120.0 (4)
N1—C8—C9	101.1 (3)		
C11—O2—C12—O3	-92.1 (4)	N1—C8—C9—C10	-83.6 (4)
C11—O2—C12—C9	24.3 (4)	N1—C8—C9—C12	27.9 (3)
N1—O3—C12—O2	94.2 (3)	C7—C8—C9—C10	154.3 (3)
N1—O3—C12—C9	-22.9 (3)	C7—C8—C9—C12	-94.2 (4)
C12—O3—N1—C8	42.7 (3)	C8—C9—C10—C11	95.7 (4)
O3—N1—C8—C7	75.1 (3)	C8—C9—C12—O2	-123.3 (3)
O3—N1—C8—C9	-43.0 (3)	C8—C9—C12—O3	-3.2 (4)
C12—O3—N1—C13	177.1 (3)	C10—C9—C12—O2	-4.2 (4)
O3—N1—C13—C14	-154.0 (3)	C10—C9—C12—O3	115.8 (3)
O3—N1—C13—C18	33.2 (4)	C12—C9—C10—C11	-15.7 (4)
C8—N1—C13—C14	-29.9 (5)	C9—C10—C11—O2	30.2 (4)
C8—N1—C13—C18	157.3 (4)	C12—O2—C11—C10	-34.7 (4)
C13—N1—C8—C7	-51.4 (5)	N1—C13—C14—C15	-172.8 (4)
C13—N1—C8—C9	-169.5 (3)	N1—C13—C18—C17	173.4 (4)

Table 3. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for (5a)

$$U_{eq} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j$$

	x	y	z	U_{eq}
O1	0.5295 (3)	0.3394 (1)	0.2706 (3)	0.060 (2)
O2	0.5307 (3)	0.5048 (1)	0.3408 (3)	0.057 (2)
O3	0.5079 (2)	0.4381 (1)	0.2114 (2)	0.048 (2)
N1	0.6422 (3)	0.4231 (1)	0.1841 (3)	0.037 (2)
C1	0.8181 (4)	0.3275 (1)	0.4736 (4)	0.045 (3)
C2	0.8690 (4)	0.2983 (2)	0.5710 (4)	0.057 (3)
C3	0.7916 (5)	0.2648 (1)	0.6273 (5)	0.065 (3)
C4	0.6627 (5)	0.2596 (1)	0.5887 (5)	0.066 (3)
C5	0.6110 (4)	0.2886 (1)	0.4926 (4)	0.055 (3)
C6	0.6881 (4)	0.3226 (1)	0.4347 (4)	0.039 (2)
C7	0.6271 (4)	0.3523 (1)	0.3309 (4)	0.039 (2)
C8	0.6849 (4)	0.4009 (1)	0.3058 (3)	0.036 (2)
C9	0.6319 (4)	0.4350 (1)	0.4109 (3)	0.041 (2)
C10	0.7228 (4)	0.4767 (1)	0.4369 (4)	0.054 (3)
C11	0.6683 (5)	0.5144 (2)	0.3490 (4)	0.060 (3)
C12	0.5118 (4)	0.4569 (1)	0.3419 (4)	0.043 (2)
C13	0.6483 (4)	0.4016 (1)	0.0588 (3)	0.034 (2)
C14	0.7464 (4)	0.3690 (1)	0.0336 (4)	0.045 (2)
C15	0.7614 (5)	0.3517 (1)	-0.0941 (4)	0.058 (3)
C16	0.6801 (5)	0.3661 (2)	-0.1958 (4)	0.061 (3)
C17	0.5836 (5)	0.3985 (2)	-0.1697 (4)	0.057 (3)
C18	0.5678 (4)	0.4164 (1)	-0.0432 (4)	0.046 (3)

Table 4. Selected geometric parameters (\AA , $^\circ$) for (5a)

O1—C7	1.223 (4)	C6—C7	1.488 (5)
O2—C12	1.397 (4)	C7—C8	1.543 (5)
O2—C11	1.431 (5)	C8—C9	1.543 (5)
O3—C12	1.423 (4)	C9—C10	1.542 (5)
O3—N1	1.460 (4)	C9—C12	1.542 (5)
N1—C13	1.408 (4)	C10—C11	1.509 (6)
N1—C8	1.450 (4)	C13—C18	1.382 (5)
C1—C6	1.388 (5)	C13—C14	1.396 (5)
C1—C2	1.394 (5)	C14—C15	1.388 (5)
C2—C3	1.371 (6)	C15—C16	1.381 (6)
C3—C4	1.377 (6)	C16—C17	1.382 (6)
C4—C5	1.384 (6)	C17—C18	1.384 (5)
C5—C6	1.386 (5)		

Data collection: COLLECT UCLA (UCLA Crystallographic package, 1984) for (4a); MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1988) for (5a). Cell refinement: LEAST UCLA for (4a); LSQUARES MSC/AFC Diffractometer Control Software for (5a). Data reduction: REDUCE UCLA for (4a); PROCESS in TEXSAN (Molecular Structure Corporation, 1989) for (5a). For both compounds, program(s) used to solve structures: SHELXS86 (Sheldrick, 1990); program(s) used to refine structures: ORFLS UCLA; molecular graphics: ORTEP (Johnson, 1965); software used to prepare material for publication: ORFFE (Busing, Martin & Levy, 1964), UCLA.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: SX1007). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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